# **Atmosphere Origin For Exoplanet Sub-Neptunes**

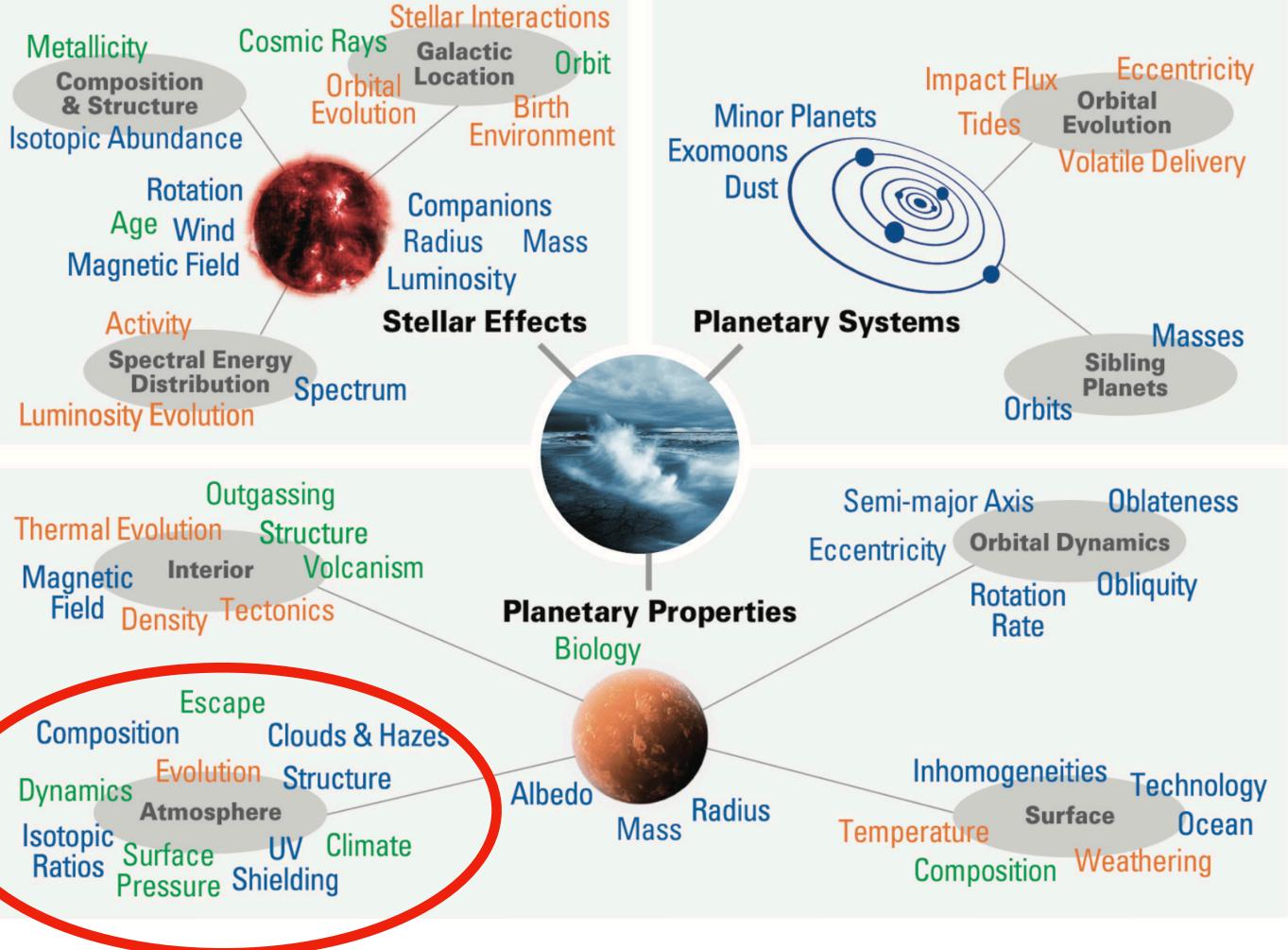
Edwin S. Kite, Bruce Fegley Jr., Laura Schaefer, and Eric B. Ford

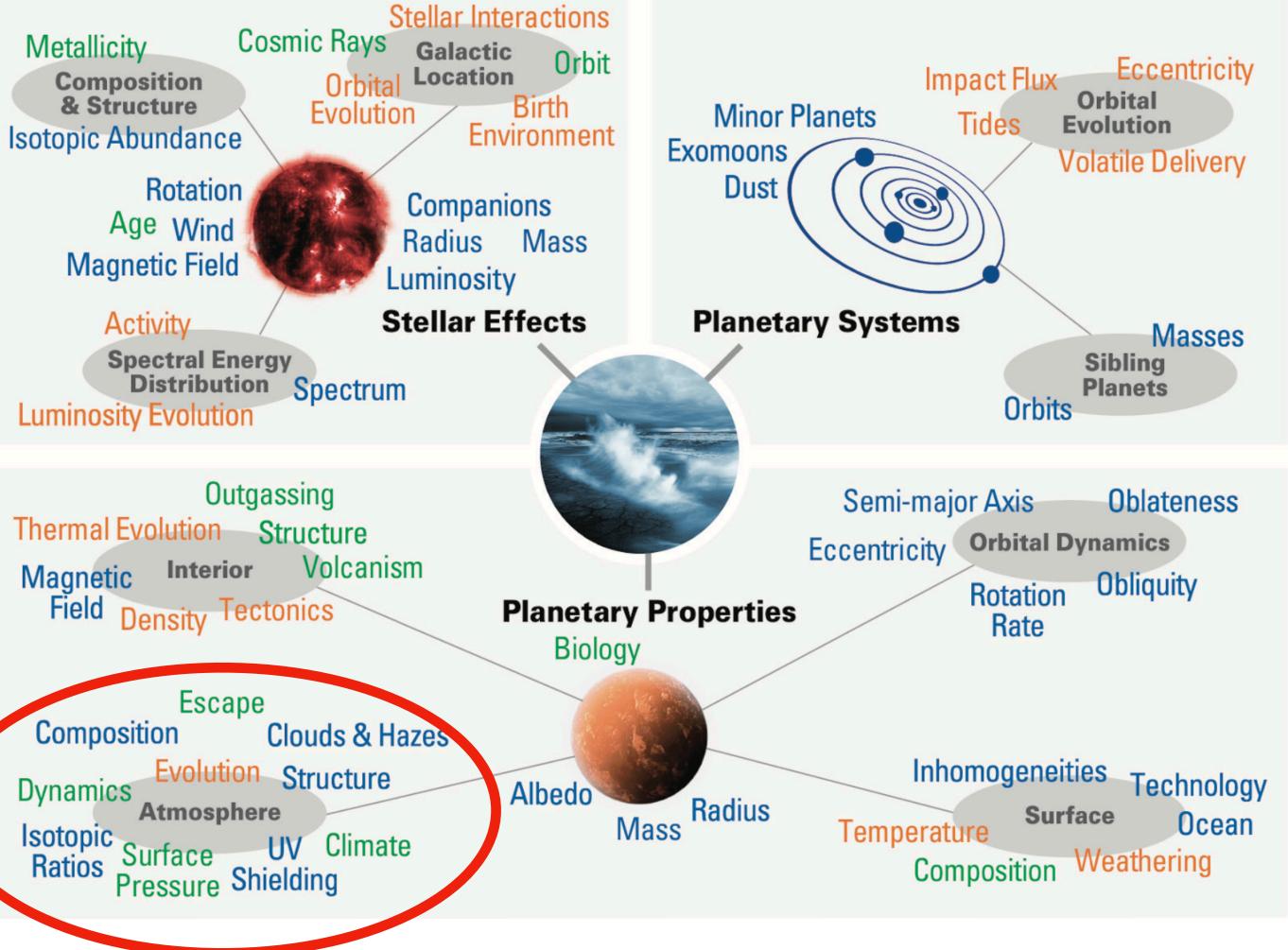
Kangning Diao Nov. 12th 2021

### Background Why atmosphere?

• Ultimate goal: find a habitat planet

 Constrain volatile delivery, planet formation...



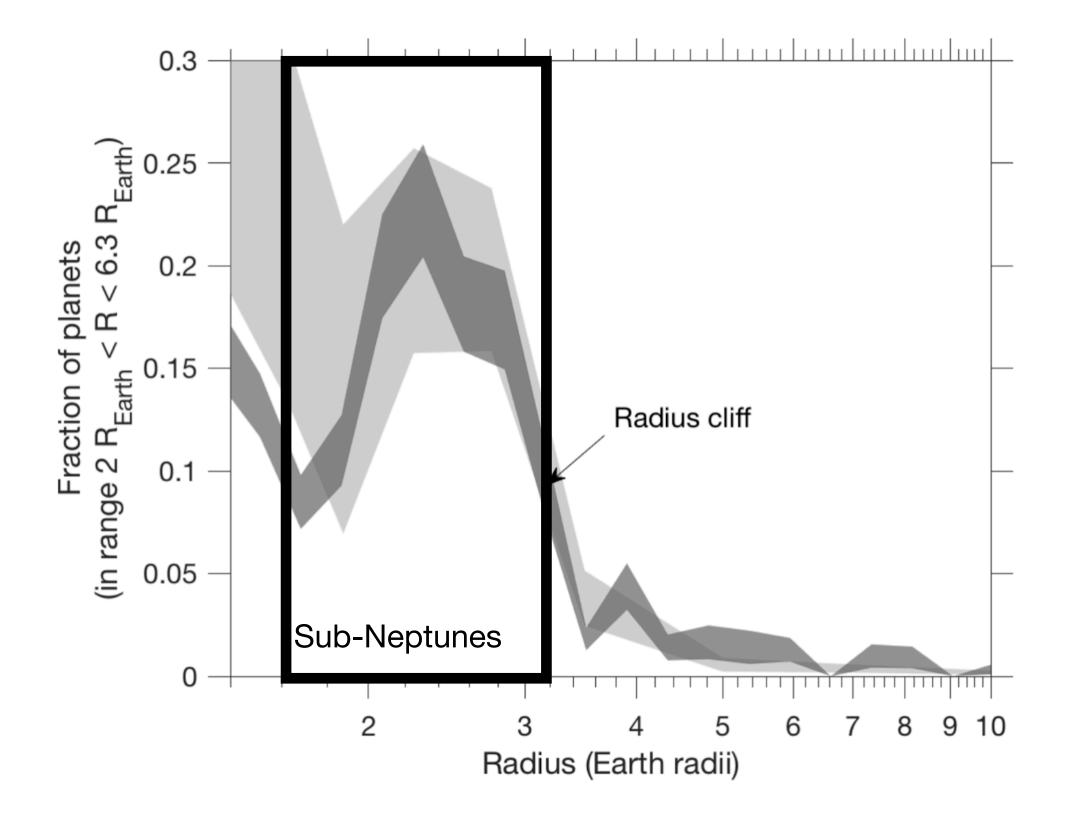


### Things that can influence the habitability

### **Background** Why sub-Neptune?

•  $1.6 - 3.2R_{\oplus}$ , density < 4g/cm<sup>3</sup>

• They are abundant



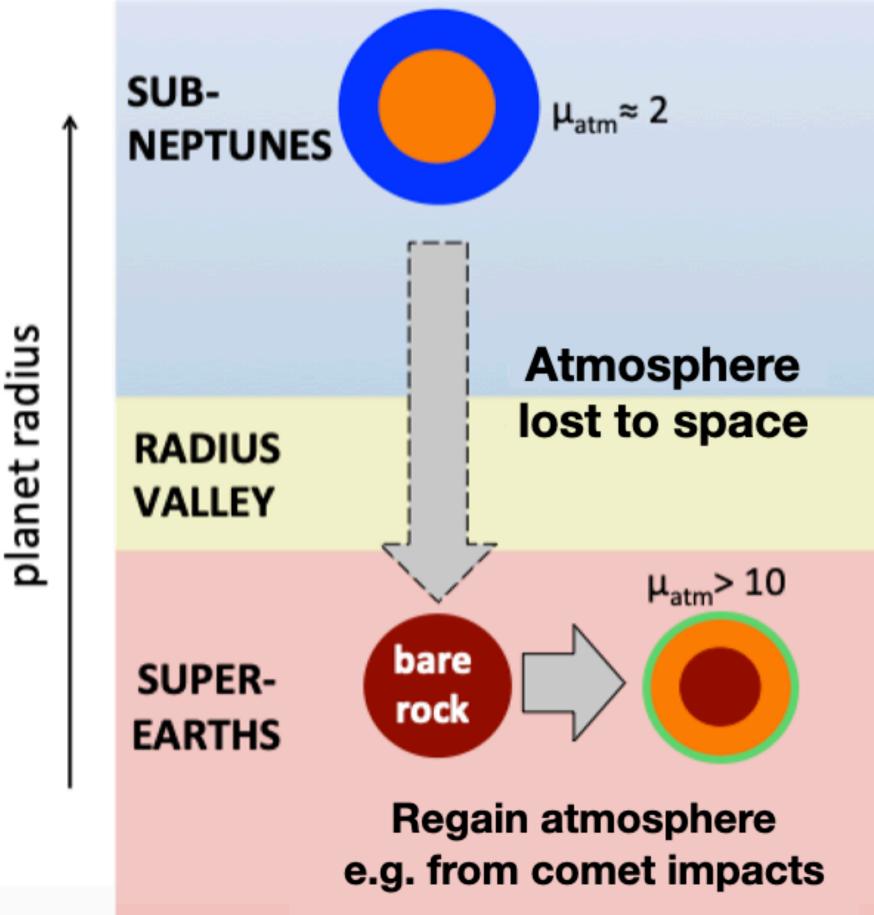
### Background Why sub-Neptune?

•  $1.6 - 3.2R_{\oplus}$ , density < 4g/cm<sup>3</sup>

• They are abundant

Super-Earth may come from them

Observations towards their atmosphere is underway

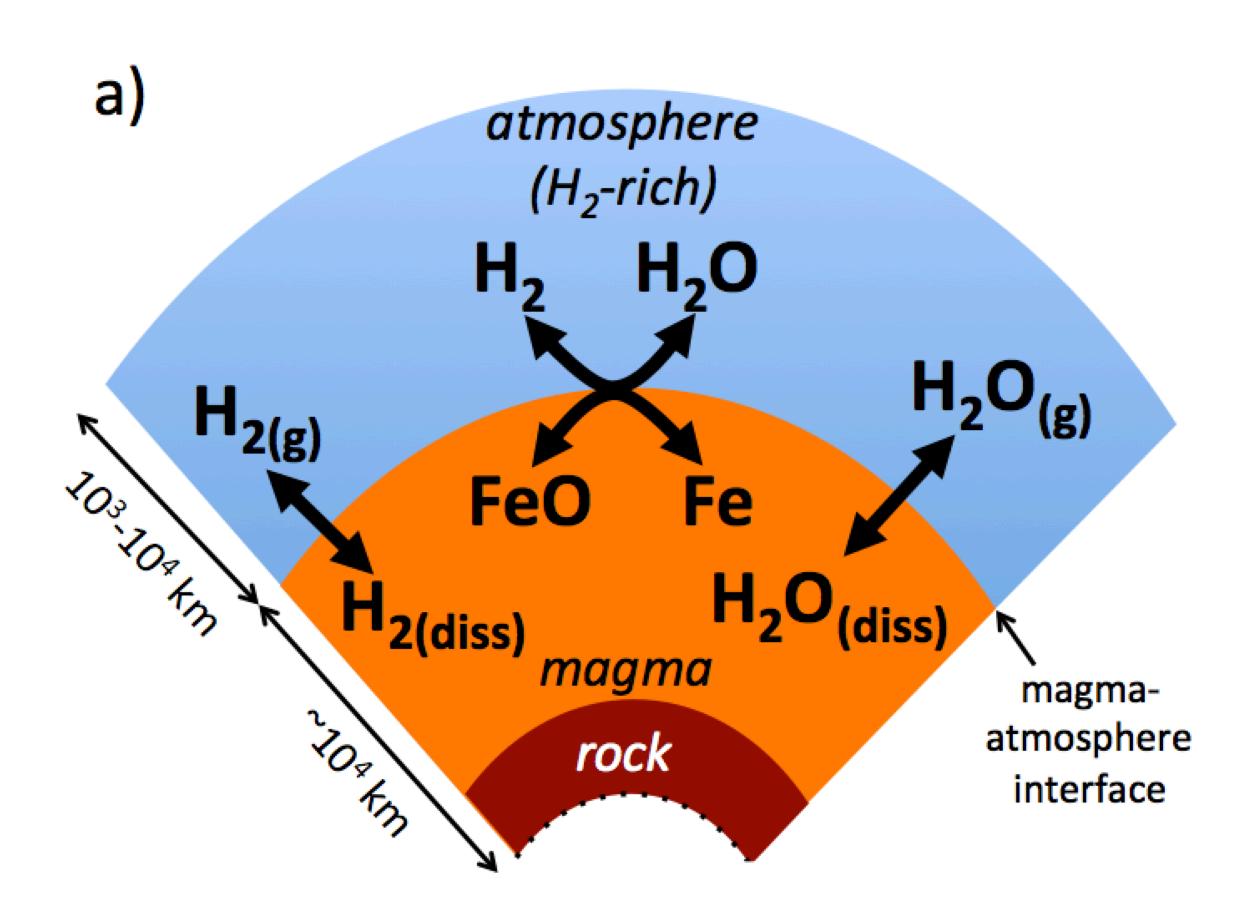


### **Background** What do we already know?

• Rock-Magma-Atmosphere

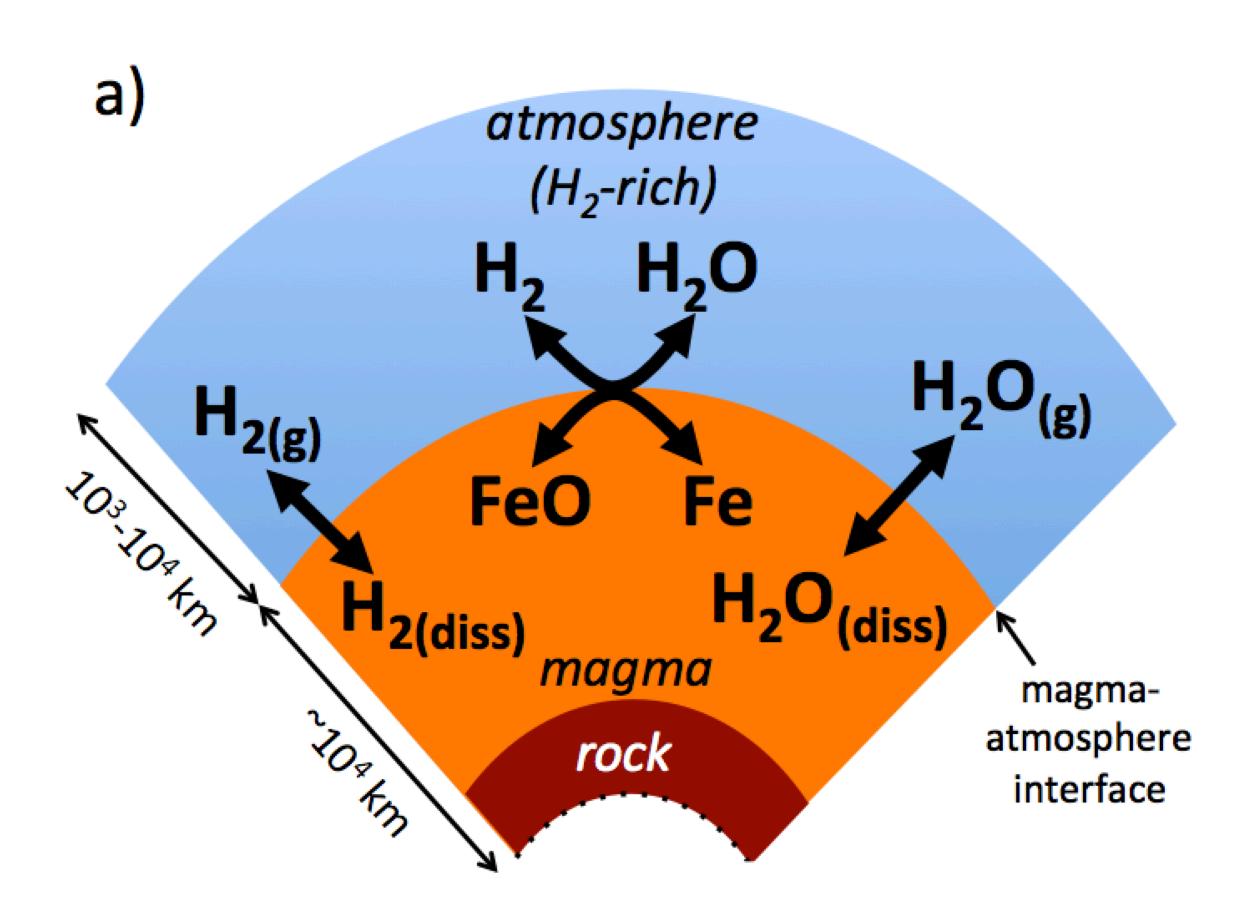
 The possible important physical and chemical reactions

 Previous work ignored the effect of these reactions on atmosphere



## Methodology Build up a model

- Contents: Mg, Fe, Si, O, H
- Reactions: Fe redox:  $Fe(liq) + H_2O(g) \iff FeO(liq) + H_2(g)$ Gas dissolution
- Environment: 3000K
  At the interface
  Strong convection



# Model parameters

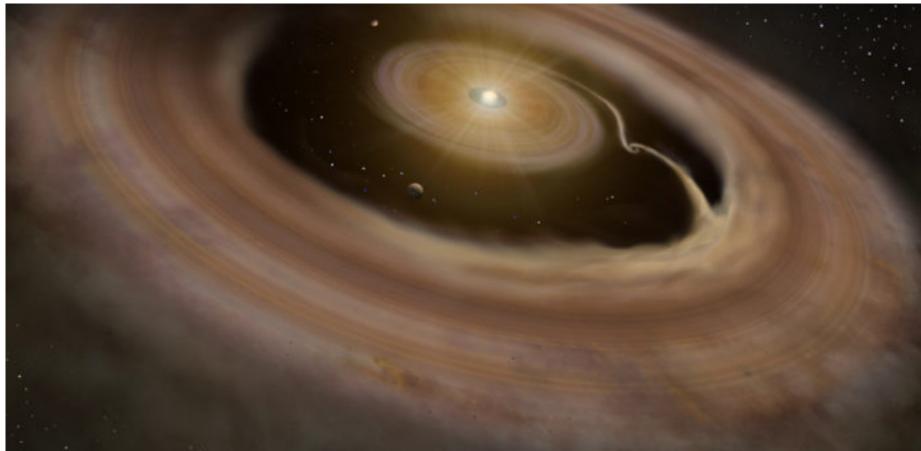
• Initial volatile:  $H_2, H_2O$ 

volatile: compounds that vaporize easier than water.

• Two ways are considered here:

accreting nebula hydrogen come with external objects like comets

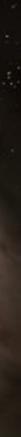
### **Accreting hydrogen**



VS

### **External water**





### Methodology **Model parameters**

Magma content: mass weight of oxidized FeO and reduced Fe

### $Fe(liq) + H_2O(g) \iff FeO(liq) + H_2(g)$

### $Fe(liq) + H_2O(g) \implies FeO(liq) + H_2(g)$

Or

# **Results**Magma matters

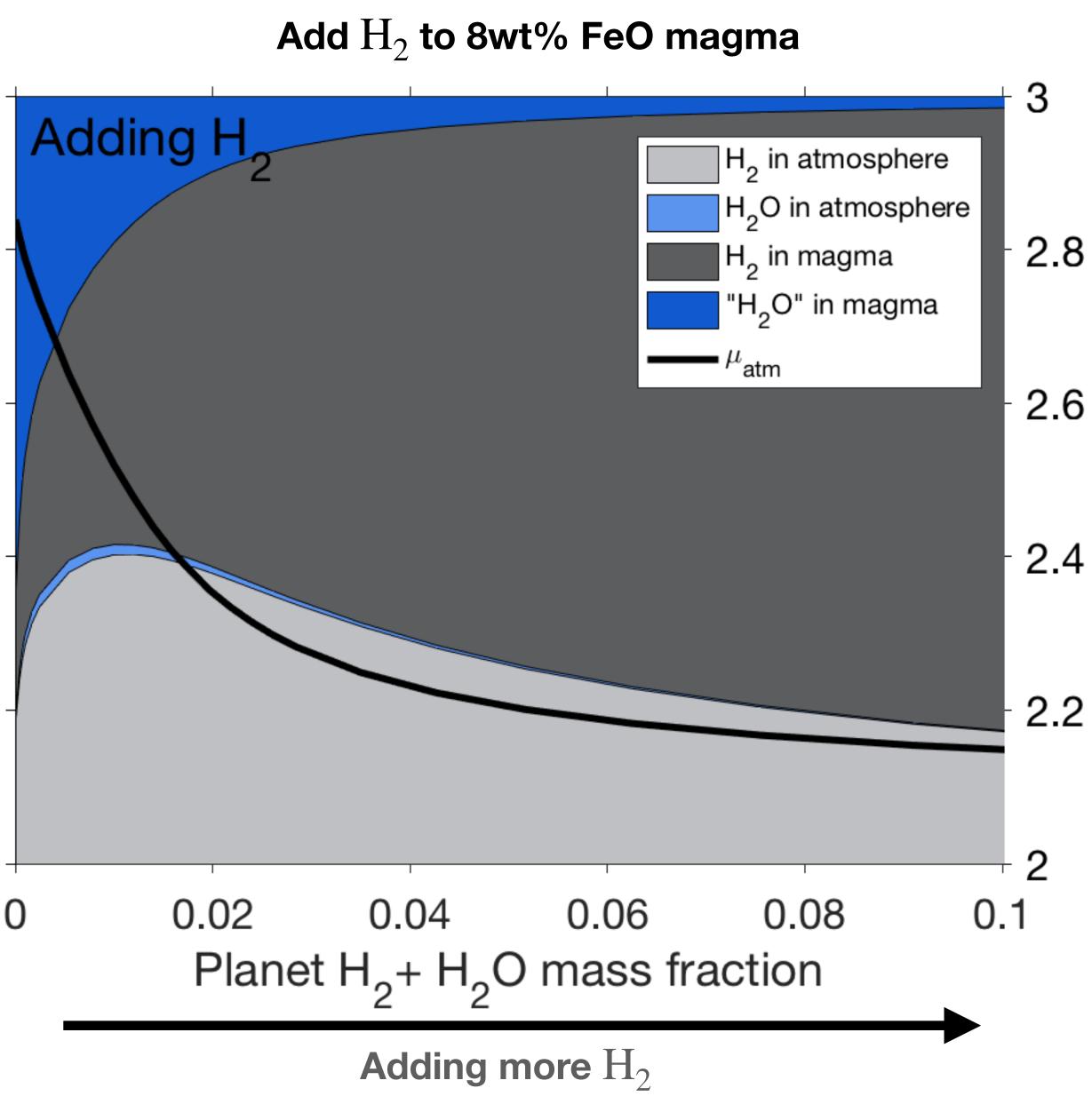
Most of gas are in the magma

 Redox reactions have a big impact Mole fraction 9.0

0.8

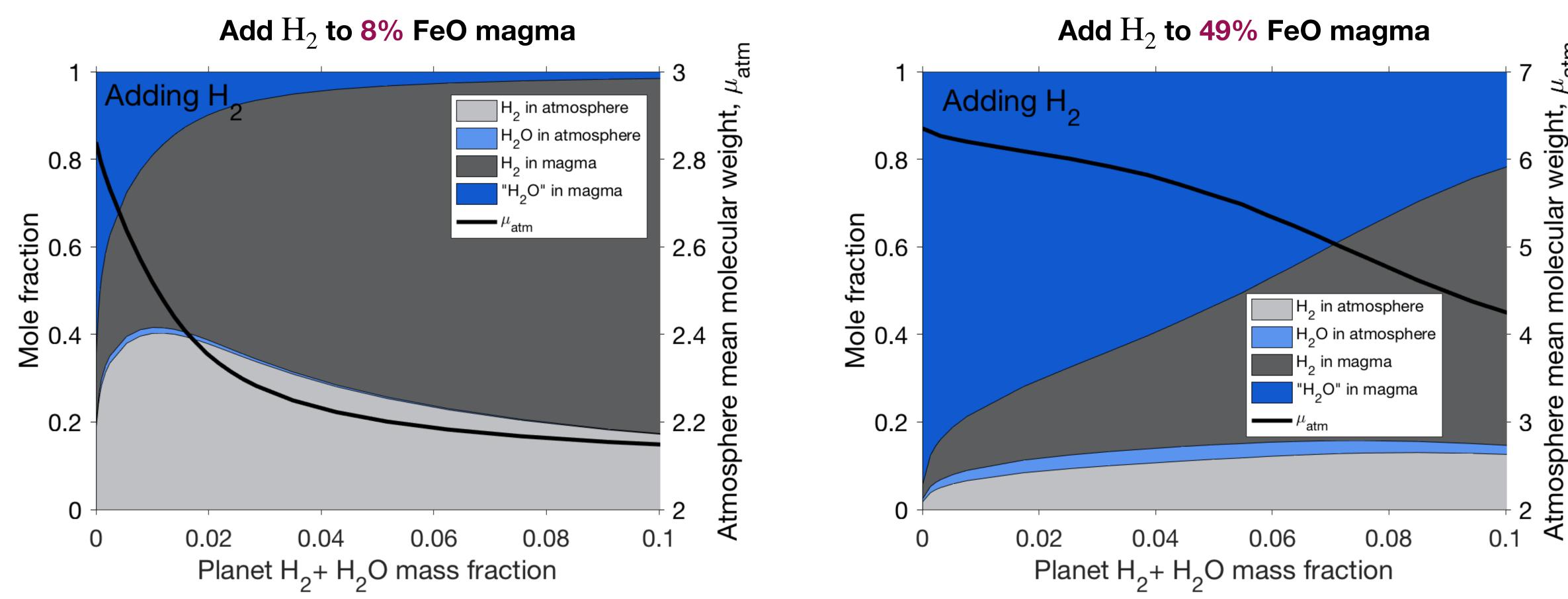
0.2

0





### Results **Magma matters**



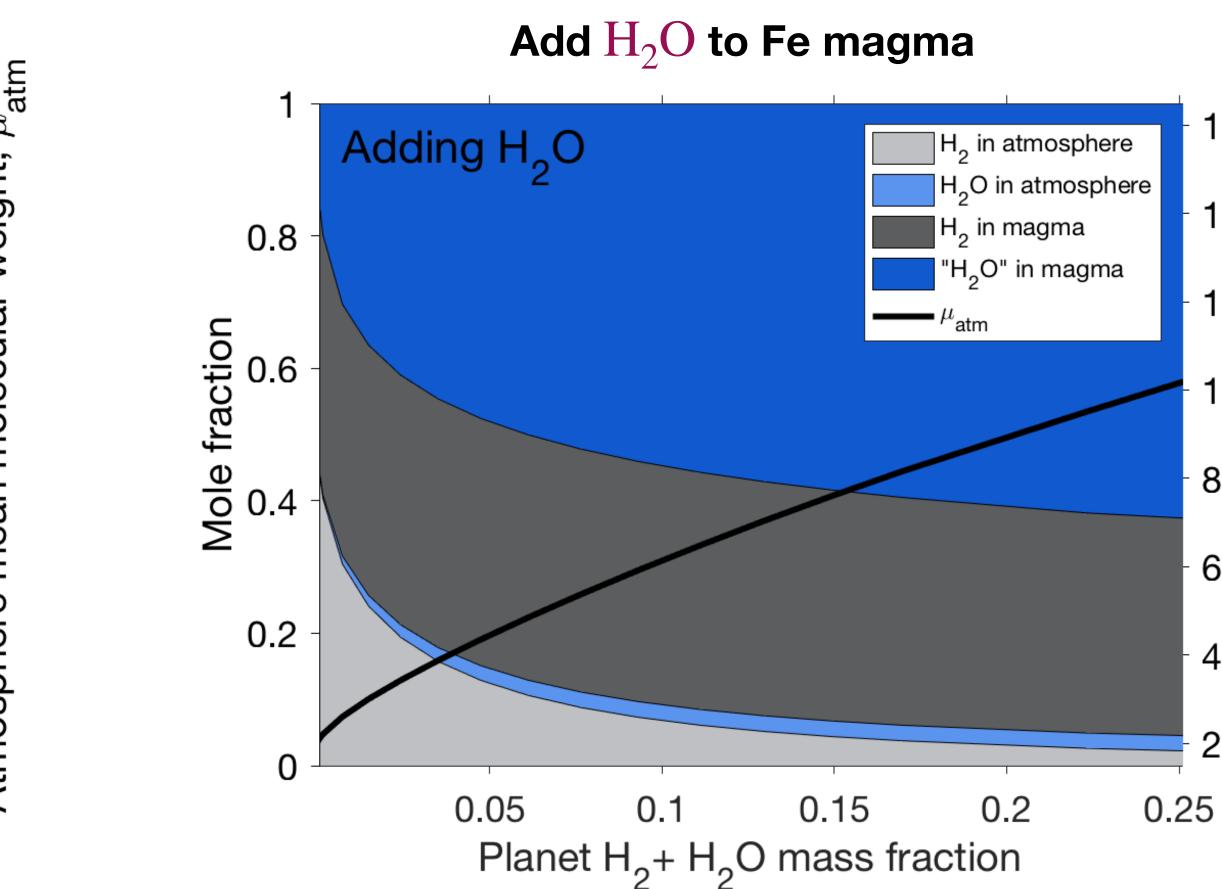
### **Redox matters!**



### Results **Magma matters**

Add  $H_2$  to 8% FeO magma mean molecular weight,  $\mu_{\rm atm}$ З Adding H H<sub>2</sub> in atmosphere H<sub>2</sub>O in atmosphere 2.8 0.8 H<sub>2</sub> in magma H<sub>2</sub>O" in magma Mole fraction 0.0 <sup>t</sup>atm 0.6 2.6 2.4 Atmosphere 2.2 0.2 0 2 0.06 0.08 0.02 0.04 0.1 0 Planet  $H_2 + H_2O$  mass fraction

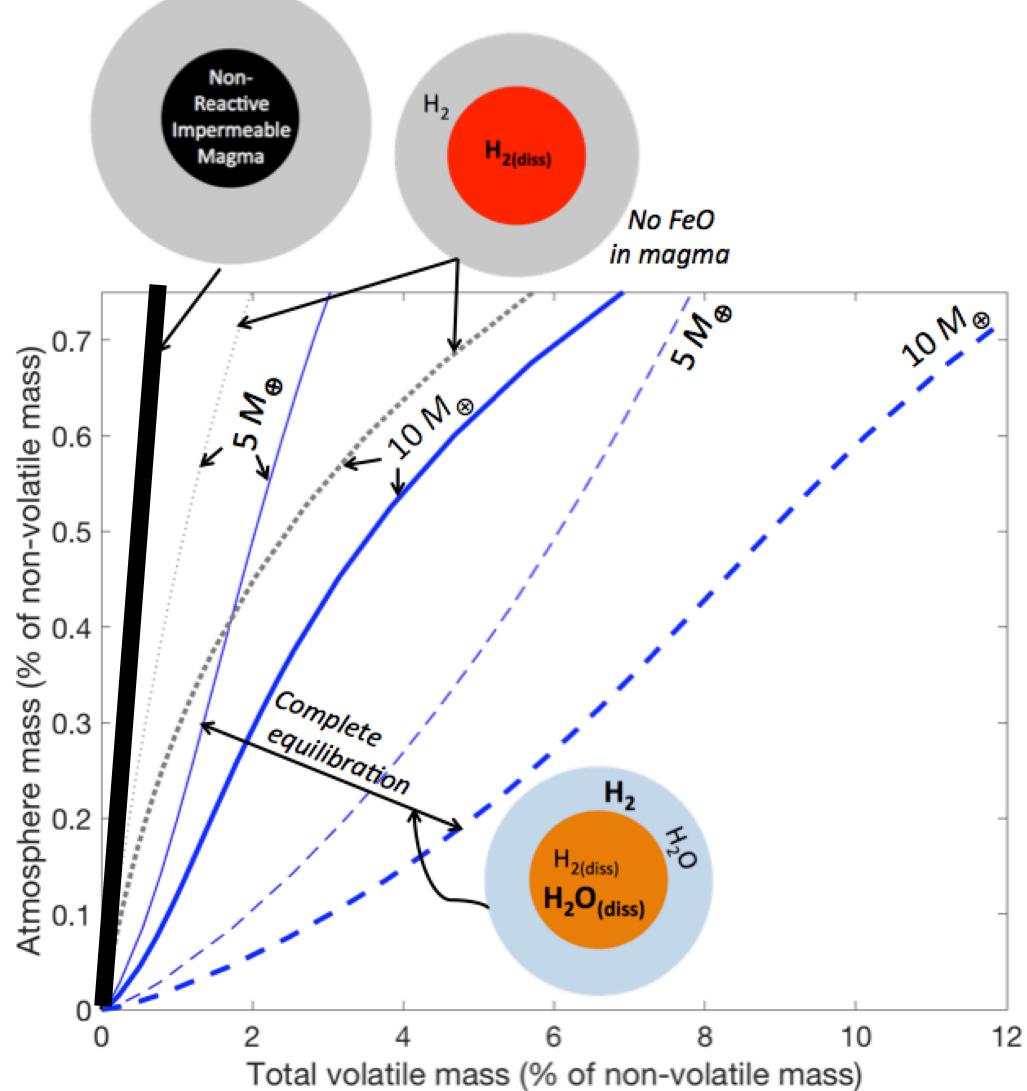
### **Initial volatile matters!**





## **Analysis** Volatile mass are variables

 Previous thought: Volatile mass can be derived from radius/atmosphere mass (Black line in the figure)



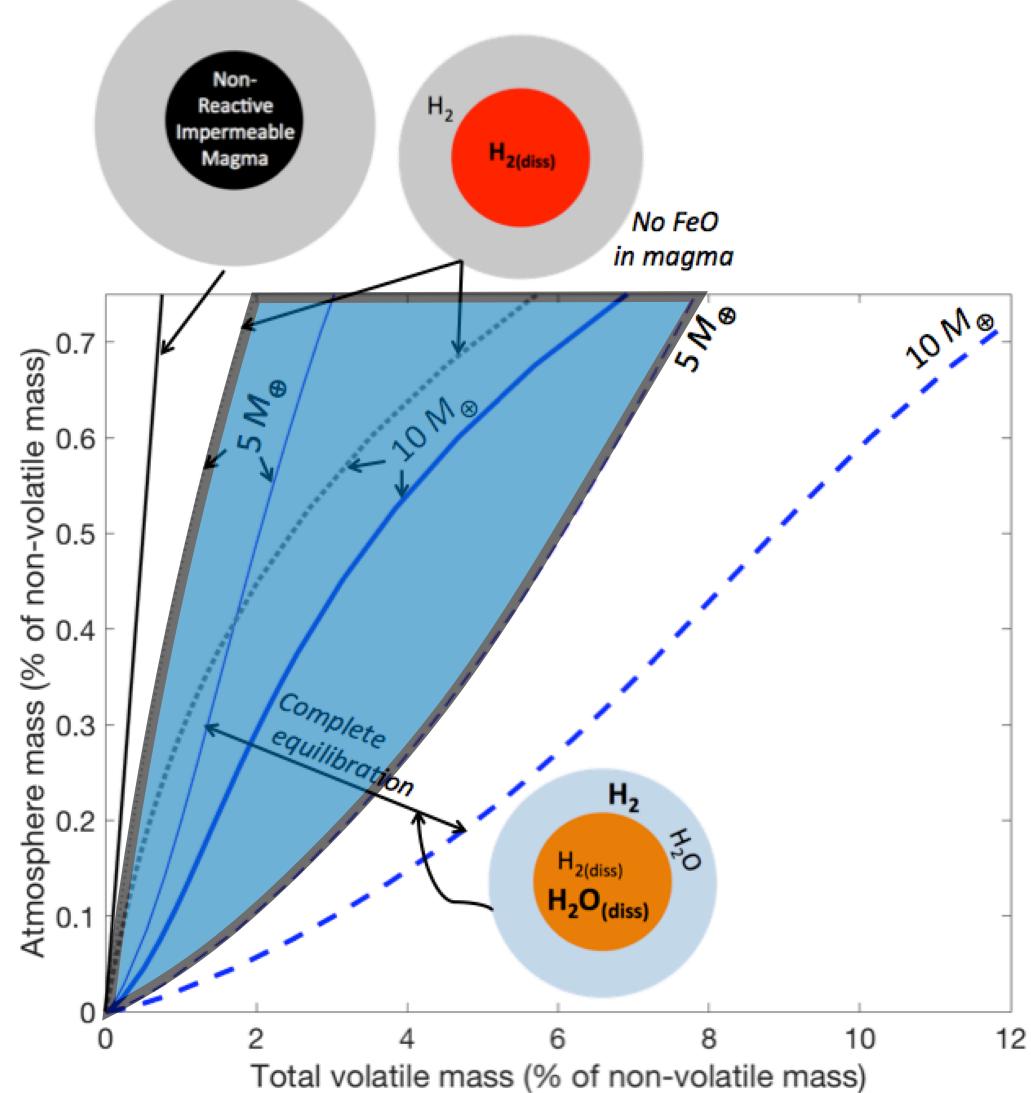
## Analysis **Volatile mass are variables**

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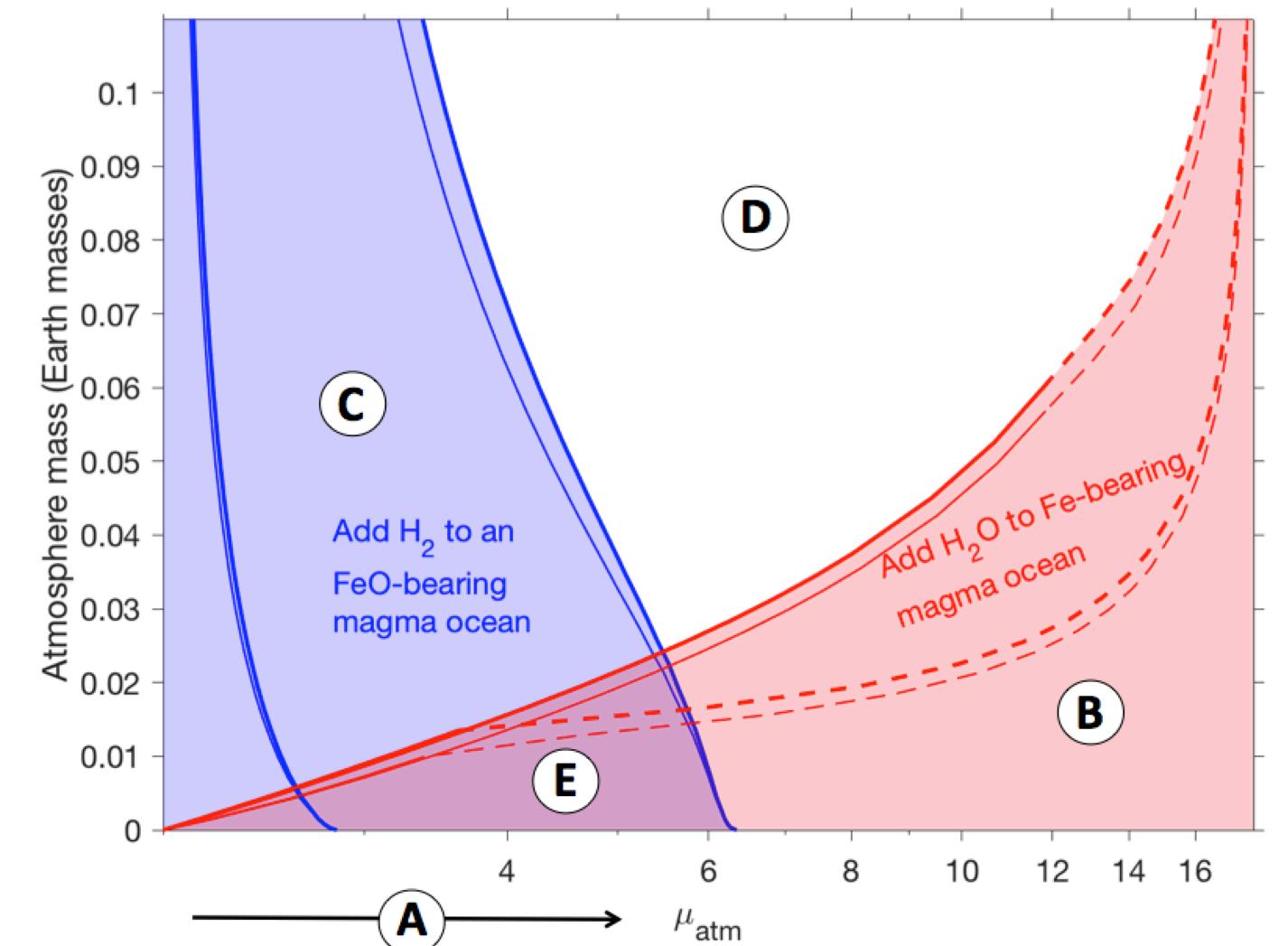
• This work:

More volatiles **Big uncertainties** 

The loss of atmosphere may be much harder



## **Analysis** Probe of atmosphere origin

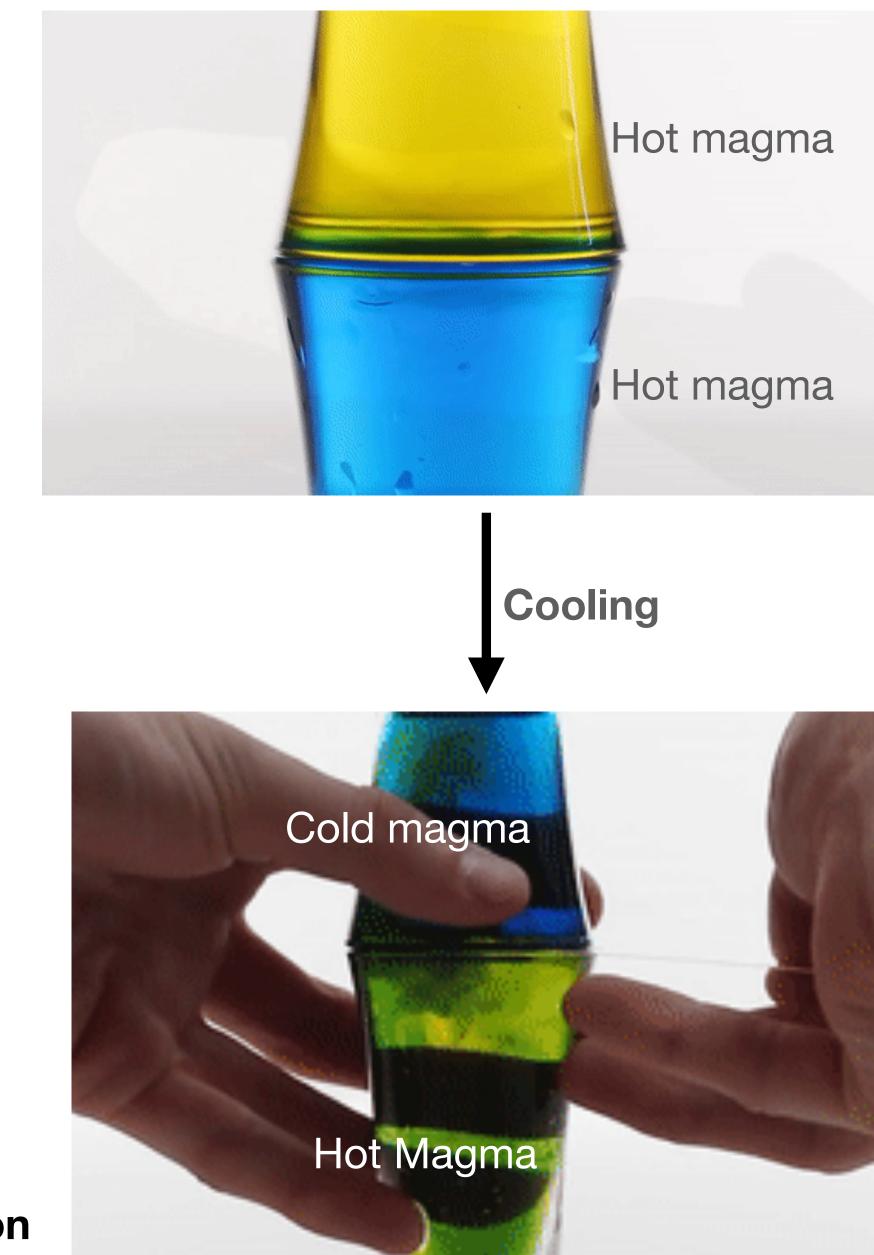


## Analysis How to test this theory?

• Previous thought: radius shrink with age

• This work: quicker shrink!

Well-stirred magma leads to complete reaction



## Conclusion Or takehome massage

• Magma matters

Observation of atmosphere can trace its origin

Magma-atmosphere interaction can be tested

# **Possible Questions**

How to measure the mean molecular weight / atmosphere mass?

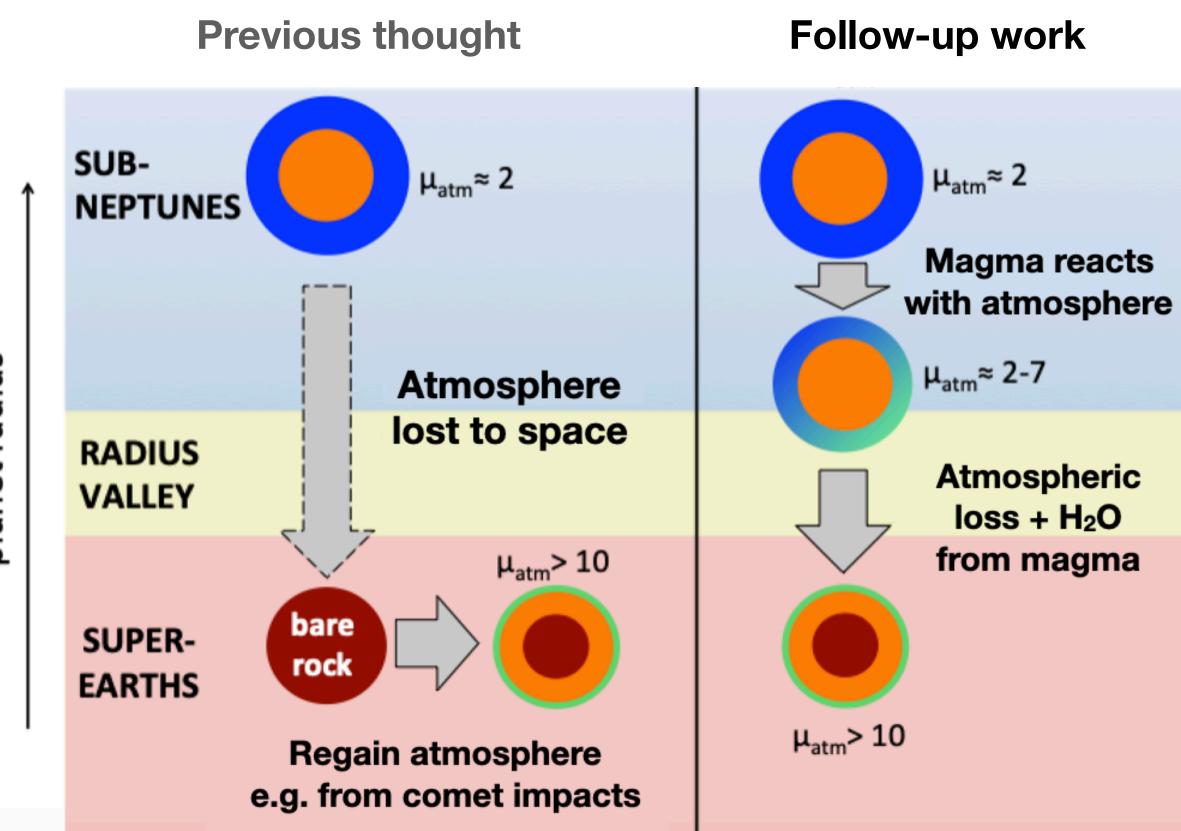
Is there any observations of sub-Neptune atmosphere? 

Can initial volatiles be both water and hydrogen?

# Discussion

- Too simplified
- Uncertainty of existing theory/ assumptions
- Reliability of chemical data

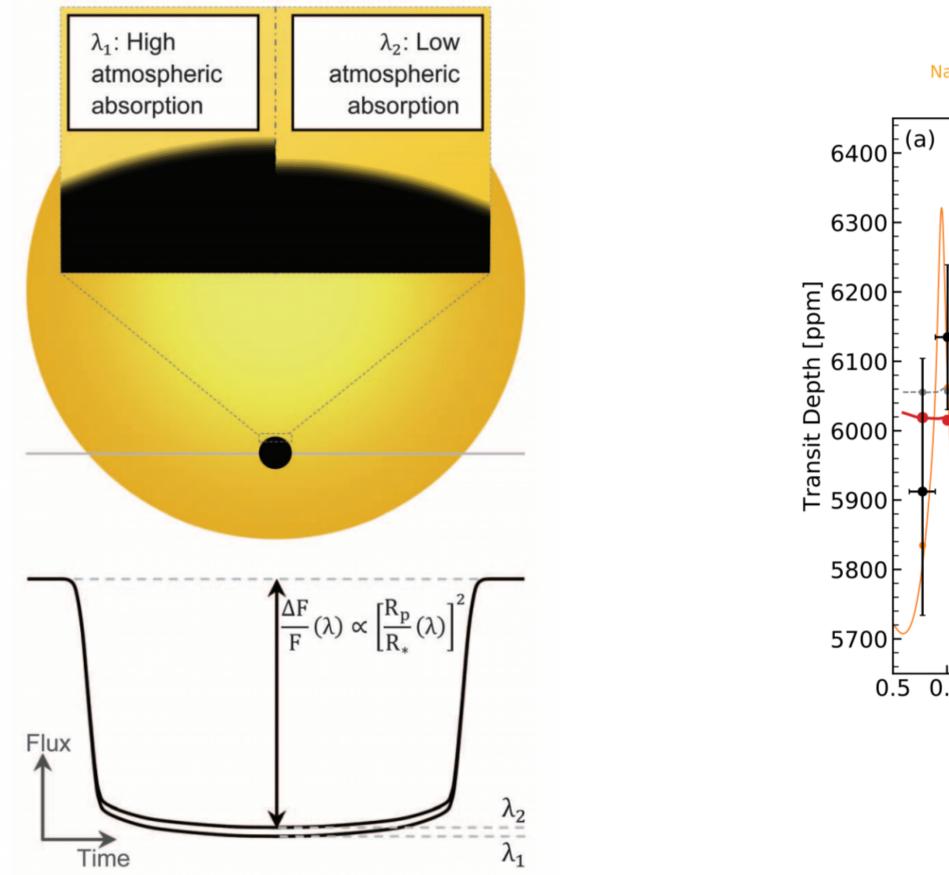
 Interesting follow-up: another way to form Super-earth



planet radius



# **Observe the atmosphere**



 $H_2O NH_3$  $H_2OCH_4$  $H_2O$ H<sub>2</sub>O NH<sub>3</sub> H<sub>2</sub>O CH<sub>4</sub> H<sub>2</sub>O  $\begin{array}{c} H_2O & H_2O & H_2O \\ H_2O/CH_4 & H_2O \\ H_2O & H_2O & H_2O \end{array} \begin{array}{c} CH_4 \\ CH_4 \\ CH_4 \end{array}$ CH<sub>4</sub> CH<sub>4</sub> CO CO<sub>2</sub> H<sub>2</sub>O CO/CO<sub>2</sub> CO/CO<sub>2</sub> CH<sub>4</sub> CH<sub>4</sub> Na  $CH_4$ (b) 6200 6150 6100 6050 6000 4.0 5.0 0.5 0.6 0.8 1.0 2.0 2.5 3.0 1.2 1.4 1.6 1.5 Wavelength[µm] Wavelength[µm]

GJ3470b

